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Research Article

Phenolic profile, antioxidant and enzyme inhibitory activity of the ethyl acetate, methanol and water extracts of *Capparis spinosa* L.

Bulent Kirkan <sup>1</sup>, Olcay Ceylan <sup>1</sup>, Cengiz Sarikurkcu <sup>1</sup>, Bektas Tepe <sup>1</sup>

Abstract: In this study, it was aimed to determine the phytochemical compositions and biological activities of ethyl acetate (EtOAc), methanol (MeOH) and water extracts obtained from the aerial parts of Capparis spinosa L. As a result of spectrophotometric analyzes, MeOH extract was found to be richer in terms of both phenolics and flavonoids compared to other extracts [81.45 mg GAEs (gallic acid equivalent)/g and 36.57 mg RE (rutin equivalent)s/g, respectively], while chromatographic analyzes showed that the extract in question contains a significant amount of hepseridin (72927.48 µg/g), quercetin (1335.88 μg/g), hyperoside (1227.73 μg/g), and 4-hydroxybenzoic acid (924.08 μg/g). Phosphomolybdenum, 2,2-diphenyl-1-picrylhydrazyl (DPPH) and 2,2-azinobis-(3-ethylbenzothiazoline-6-sulphonic acid) (ABTS) radical scavenging, Cupric Reducing Antioxidant Power (CUPRAC) and Ferric Reducing Antioxidant Power (FRAP) reducing and ferrous ion chelating activity tests resulted in superiority of MeOH extract [371.0, 44.93, 56.46, 91.77, 52.61 mg TEs (trolox equivalent)/g and 14.85 mg EDTAEs/g, respectively]. On the other hand, EtOAc extract exhibited higher activity than other extracts in acetylcholinesterase (AChE), butyrylcholinesterase (BChE), α-amylase, and α-glucosidase inhibitory activity tests [3.29, 2.12 mg GALAEs (galanthamine equivalent)/g, 541.01 and 1584.20 mg ACEs (acarbose equivalent)/g, respectively]. The tyrosinase inhibitory activity test resulted in the superiority of MeOH extract [41.90 mg KAEs (kojic acid equivalent)/g]. A strong correlation was determined between the phenolic and flavonoid contents of the extracts and their antioxidant activities.

## ARTICLE HISTORY

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#### **KEYWORDS**

Capparis spinosa, LC-ESI-MS/MS, Antioxidant, Enzyme inhibitory, Chemical composition,

# 1. INTRODUCTION

Plants can be used in industries such as medicine, pharmacy, food, cosmetics, etc., due to their pharmacological/biologically active phytochemicals, and therefore, new plants are attracting the attention of researchers every day (Orphanides *et al.*, 2016). Researchers have revealed that many phytochemicals such as polyphenols, flavonoids, flavonoids, etc. can be used as critical functional compounds in the treatment of many metabolic diseases (Ng *et al.*, 2012). Since

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<sup>&</sup>lt;sup>1</sup>Suleyman Demirel University, Water Institute, TR-32260, Isparta, TURKEY

<sup>&</sup>lt;sup>2</sup>Mugla Sitki Kocman University, Faculty of Science, TR-48000, Mugla, TURKEY

<sup>&</sup>lt;sup>3</sup>Afyonkarahisar Health Sciences University, Faculty of Pharmacy, TR-03100, Afyonkarahisar, TURKEY

<sup>&</sup>lt;sup>4</sup>Kilis 7 Aralik University, Faculty of Science and Literature, TR-79000, Kilis, Turkey

<sup>\*</sup>CONTACT: Bektas TEPE bektastepe@yahoo.com Kilis 7 Aralik University, Faculty of Science and Literature, Department of Molecular Biology and Genetics, Kilis, Turkey

plants have low cost and are sustainable sources of phytochemicals, it is of great importance to use those with proven biological/pharmacological activities for industrial use as an alternative to synthetic compounds (Samy *et al.*, 2005; Ng *et al.*, 2020).

Foods deteriorate over time due to lipid oxidation and lose their quality (Yanishlieva & Marinova, 2001). The deterioration of foods with high lipid content can be delayed by the addition of compounds that prevent oxidation during processing. The most effective way to control lipid oxidation is to benefit from antioxidant compounds (Shahidi & Zhong, 2015). These compounds can be found widely in plants and animals, as well as chemically synthesized. Polyphenols and tocopherols are among the most potent antioxidant compounds and are abundant in many vegetables, fruits and grains. In addition, there are various reports of compounds with antioxidant effects in fish, algae and shellfish (Shahidi & Amarowicz, 1996; Amarowicz *et al.*, 1999; Athukorala *et al.*, 2003). In the last decades, researchers have focused on many plant species for the discovery of new and more effective natural compounds that can be used instead of synthetic antioxidants to extend the shelf life of foods. As a result of these studies, many antioxidant phytochemicals have been identified (Liyana-Pathirana *et al.*, 2006; Shahidi & Zhong, 2007; Cumby *et al.*, 2008).

Plants are of particular interest to researchers because they contain phytochemicals with enzyme inhibitory activity as well as antioxidant activities. In the enzyme inhibitory activity studies intensified in recent years, it has been reported that some plant species or some phytochemicals found in these species exhibit inhibitory activity such as cholinesterase (Hung et al., 2008; Loizzo et al., 2010; Pinho et al., 2013), α-amylase/α-glucosidase (Liu et al., 2017; Rasouli et al., 2017; Tan et al., 2017), tyrosinase (Kubo & Kinst-Hori, 1999; Likhitwitayawuid, 2008; Maisuthisakul & Gordon, 2009), etc.

Capparis spinosa L. is an industrial plant species distributed in western and central Asia and along the Mediterranean coastline (Trombetta et al., 2005; Rahimi et al., 2020). This herb has been traditionally used by people for many years in the treatment of various diseases (gout, rheumatism, etc.) (Romeo et al., 2007; Aliyazicioglu et al., 2013; Zhang et al., 2018). Local people living in countries bordering the Mediterranean coastline also frequently benefit from C. spinosa's analgesic properties (Fu et al., 2008). In studies conducted by researchers, it has been reported that the aerial parts, roots or seeds of the plant exhibit many biological/pharmacological activities (anti-allergic, immunomodulatory, anti-inflammatory, antimicrobial, anti-histaminic, antiviral, etc.) (Trombetta et al., 2005; Tlili et al., 2011; Kulisic-Bilusic et al., 2012).

The aim of this study was to determine the chemical compositions of ethyl acetate (EtOAc), methanol (MeOH) and water extracts obtained from C. spinosa by qualitative and quantitative chromatographic methods, in vitro antioxidant and to document their inhibitory activities on acetylcholinesterase (AChE), butyrylcholinesterase (BChE),  $\alpha$ -amylase,  $\alpha$ -glucosidase, and tyrosinase.

#### 2. MATERIAL and METHODS

# 2.1. Plant Material and Extract Preparation

Aerial parts of *C. spinosa* was collected from Camlibel village, Kavaklidere, Mugla-Turkey (780 m., 37° 24 849'N 28° 27 688'E) (Herbarium number: O.1196). Dr. Olcay Ceylan (Mugla Sitki Kocman University) performed the identification of the plant material. Aerial parts of the plants were used as the study material to obtain EtOAc, MeOH and water extracts [extract yields: 10.20, 11.31 and 21.58% (w/w), respectively]. Details of the extraction procedure can be found in supplementary file.

### 2.2. Determination of the Phenolic Compositions of the Extracts

Details of the spectrophotometric and chromatographic analysis were given in supplementary file (Zengin *et al.*, 2017; Cittan & Çelik, 2018).

# 2.3. Biological Activity

Details of the antioxidant (Apak et al., 2006; Tepe et al., 2011; Kocak et al., 2016; Zengin et al., 2017; Sarikurkcu et al., 2020) and enzyme inhibitory activity (Ozer et al., 2018) tests were given in supplementary file.

## 2.3. Statistical Analysis

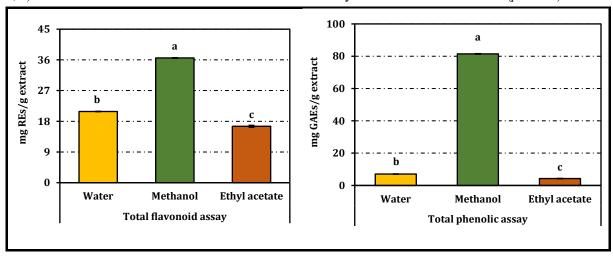
Details of the relative antioxidant capacity index (RACI) (Sun & Tanumihardjo, 2007) and statistical analysis can be found in the supplementary file.

#### 3. RESULTS / FINDINGS

## 3.1. Chemical Composition

The total amounts of phenolic and flavonoid compounds of the extracts are given in Figure 1. According to the data obtained by spectrophotometric method, MeOH extract was found to be rich in both phenolics and flavonoids. The total amount of phenolic and flavonoid compounds of this extract was 81.45 mg GALAEs/g and 36.57 mg REs/g, respectively. Although the amounts of phenolic and flavonoid compounds of EtOAc and water extracts were close to each other, the chemical compositions of these extracts were statistically different from each other (p < 0.05).

**Figure 1.** Amounts of total flavonoids and phenolics in the extracts of *C. spinosa*. Different letters (a, b, c) on the bars show that the relevant data are statistically different from each other (p < 0.05).



The chemical composition data of the extracts obtained by chromatographic methods are given in Table 1. According to the data in the table, it was understood that none of the extracts contained (+)-catechin, luteolin 7-glucoside, apigenin 7-glucoside, pinoresinol, kaempferol and luteolin. Chromatographic analyses showed that hepseridine was present in high amounts in the MeOH extract (72927.48  $\mu$ g/g). Quercetin (1335.88  $\mu$ g/g), hyperoside (1227.73  $\mu$ g/g), 4-hydroxybenzoic acid (924.08  $\mu$ g/g) were also found in the MeOH extract. In addition to these phytochemicals, *p*-coumaric acid was also found in high amounts in EtOAc and water extracts.

### 3.2. Antioxidant Activity

The total antioxidant activities, reducing powers, radical scavenging and chelating capacities of the extracts are given in Figure 2. On the figures, the statistical relationship between the antioxidant activities of the extracts and each other is also indicated with small letters.

In all antioxidant test systems presented in Figure 2, the MeOH extract exhibited significantly higher activity than the others. The activity value of this extract in phosphomolybdenum, DPPH and ABTS radical scavenging, CUPRAC and FRAP reducing and ferrous ion chelating assays were 371.0, 44.93, 56.46, 91.77, 52.61 mg TEs/g and 14.85 mg EDTAEs/g, respectively. In the radical scavenging and ferrous ion chelating activity tests, the MeOH extract was followed by the water extract (18.80, 35.36 mg TEs/g and 9.97 mg EDTAEs/g, respectively), while the EtOAc extract ranked second in the phosphomolybdenum and CUPRAC reducing assays (329.40 and 48.34 mg TEs/g, respectively). In the FRAP reducing assay, however, no statistically significant difference was found between the antioxidant activities of water and EtOAc extracts.

**Table 1.** Concentrations of selected phenolic compounds in the extracts of *C. spinosa* (μg/g extract).

Compound	EtOAc	MeOH	Water	
Gallic acid	$3.84 \pm 0.02^b$	$8.46{\pm}0.40^a$	$3.80\pm0.10^{b}$	
Protocatechuic acid	$646.52 \pm 0.37^a$	$179.25\pm2.86^{b}$	$13.36 \pm 0.29^{c}$	
3,4-Dihydroxyphenylacetic acid	$13.36 \pm 0.43^a$	$14.28 \pm 0.37^a$	$14.76 \pm 0.19^a$	
Pyrocatechol	$26.39 \pm 0.14^{b}$	$42.93\pm2.92^a$	$33.11\pm1.04^{b}$	
(+)-Catechin	nd	nd	nd	
Chlorogenic acid	$4.33\pm0.01^{a}$	$4.06 \pm 0.10^b$	$4.19\pm0.01^{ab}$	
(-)-Epicatechin	$2.61 \pm 0.09^{b}$	$3.08\pm0.13^{a}$	$2.41 \pm 0.02^{b}$	
2,5-Dihydroxybenzoic acid	$16.93\pm0.62^a$	$11.75 \pm 0.90^b$	$14.51\pm1.32^{ab}$	
4-Hydroxybenzoic acid	$648.78 \pm 1.04^{b}$	$924.08\pm2.26^a$	$106.19\pm3.28^{c}$	
Vanillic acid	$215.30\pm17.44^{b}$	$305.77 \pm 14.43^a$	$138.24\pm5.15^{c}$	
Caffeic acid	$12.91 \pm 0.85^{b}$	$24.96 \pm 1.84^a$	$14.13 \pm 0.43^b$	
Syringic acid	$9.74{\pm}0.06^{b}$	$53.98 \pm 6.66^{aa}$	$5.65\pm0.10^{b}$	
3-Hydroxybenzoic acid	$10.64\pm0.60^{a}$	$11.74\pm0.77^a$	$11.39\pm0.23^a$	
Vanillin	$7.85 \pm 0.22^{c}$	$19.42\pm2.16^{b}$	$31.79\pm1.21^a$	
Verbascoside	$5.78 \pm 0.08^a$	$6.04\pm0.13^a$	$5.90\pm0.03^{a}$	
Taxifolin	$7.11\pm0.16^{c}$	$15.60\pm0.11^a$	$8.77 \pm 0.26^b$	
p-Coumaric acid	$144.54 \pm 0.56^b$	$693.59\pm10.15^a$	$34.32\pm2.24^{c}$	
Sinapic acid	$4.97 \pm 0.13^{c}$	$33.80\pm0.14^a$	$6.06 \pm 0.04^b$	
Ferulic acid	$71.80 \pm 1.21^b$	$176.20\pm4.65^a$	$13.39\pm2.38^{c}$	
Luteolin 7-glucoside	nd	nd	nd	
Hyperoside	$4.53\pm0.19^{b}$	$1227.73\pm16.22^a$	$9.32 \pm 0.29^b$	
Hesperidin	$260.27 \pm 3.47^a$	$72927.48\pm659.21^{b}$	$155.34\pm1.26^a$	
Rosmarinic acid	$16.83 \pm 1.01^b$	$28.82 \pm 0.63^a$	$25.49 \pm 1.89^a$	
Apigenin 7-glucoside	nd	nd	nd	
2-Hydroxycinnamic acid	$3.02 \pm 0.09^b$	$2.06 \pm 0.07^{c}$	$3.54\pm0.05^{a}$	
Eriodictyol	$9.56 \pm 0.12^b$	$46.83 \pm 2.88^a$	$13.64 \pm 0.11^{b}$	
Pinoresinol	nd	nd	nd	
Quercetin	$5.42 \pm 0.15^{c}$	$1335.88 \pm 6.51^a$	$100.11 \pm 1.39^b$	
Kaempferol	nd	nd	nd	
Luteolin	nd	nd	nd	

Different letters (a, b, c) within the same row show that the relevant data are statistically different from each other (p < 0.05). nd: Not detected.

The relative antioxidant capacity index (RACI) data (Figure 3), in which extracts were compared with each other according to their activity potentials, taking into account all the activities obtained from the antioxidant activity tests, confirmed the data obtained from the antioxidant activity tests. According to the data presented in the figure, the MeOH extract ranked first with a RACI value of 1.20. It was followed by water and EtOAc extracts with RACI values of -0.55 and -0.62, respectively.

**Figure 2.** Antioxidative capacity of the extracts of *C. spinosa*. Different letters (a, b, c) on the bars show that the relevant data are statistically different from each other (p < 0.05).

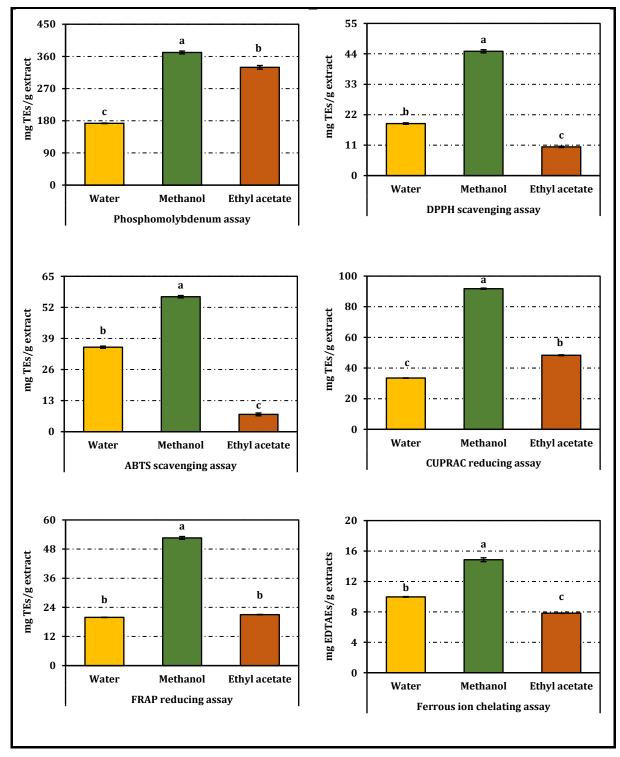
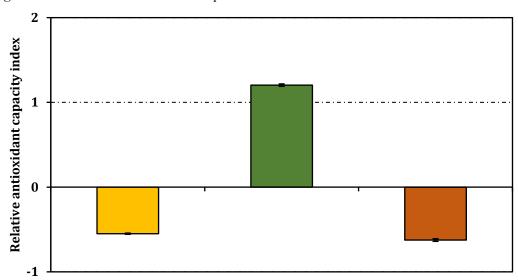


Figure 4 shows the correlation between the antioxidant activities of the extracts and their RACI values. A high correlation was found between the antioxidant activities of the extracts and the RACI values in all tests except the phosphomolybdenum assay. However, in the phosphomolybdenum test, the correlation between the total antioxidant activity of the EtOAc extract and the RACI value was found to be lower than in the other test systems.



**Figure 3.** RACI of the extracts of *C. spinosa*.

# 3.3. Enzyme Inhibitory Activity

Water

Figure 5 shows the inhibitory activity potentials of C. spinosa extracts on AChE, BChE,  $\alpha$ -amylase,  $\alpha$ -glucosidase and tyrosinase.

Methanol

Ethyl acetate

As can be seen from Figure 5, EtOAc extract exhibited higher inhibitory activity than the others in all test systems except the tyrosinase inhibitor activity test. The inhibitory activities of the extract in question in AChE, BChE,  $\alpha$ -amylase and  $\alpha$ -glucosidase tests were 3.29, 2.12 mg GALAEs/g, 541.01, and 1584.20 mg ACEs/g, respectively. The tyrosinase inhibitory activity test resulted in the superiority of MeOH extract (41.90 mg KAEs/g). In this assay, no statistical difference was found between the activity potentials of the water and EtOAc extracts. While water and MeOH extracts were not active in the BChE inhibitory activity tests, the water extract remained inactive in the  $\alpha$ -glucosidase inhibitor activity assay.

### 3.4. Correlation Coefficients

Table 2 shows the correlation between the biological activity data of the extracts in the tests given above and their chemical compositions.

According to the correlation coefficients given in Table 2, there was a strong correlation between the phenolic and flavonoid contents of the extracts and their antioxidant activities (correlation coefficients were above 0.9). The relationship between these compounds and tyrosinase inhibitory activity was also found to be high. In addition, the relationship between protocatechuic acid, 4-hydroxybenzoic acid, vanillic acid, p-coumaric acid, ferulic acid, hesperidin, hyperoside, and quercetin contents of the extracts and their antioxidant and tyrosinase inhibitor activities were also found statistically significant.

1,5 1,4 1,5 1,0 1,2 0,7 errous ion chelating 1,0 0,6 0,5 0,0 RACI 0,0 0,0 0,0 -0,6 -0.5 -1,2 -1,8 -1,0 Water Methanol Ethyl acetate Water Methanol Ethyl acetate 1,4 1,2 1,5 1,5 **JPPH scavenging** 1,0 1,0 0,0 0,5 0,5 0,0 0,0 -0,5 -0,5 -1,0 -2,1 -1.0 -1.8 Water Methanol Ethyl acetate Water Methanol Ethyl acetate 1,5 1,4 1,5 1,4 CUPRAC reducing 0,7 RAP reducing 1.0 1,0 0,5 0,5 RACI 0,0 0,0 -0,5 -0,5 -1,0 -2.1 -1,0 Water Methanol Ethyl acetate Water Methanol Ethyl acetate

**Figure 4.** Antioxidant activity (solid dark blue line with circle) and RACI (dashed red line with triangle) of the extracts of *C. spinosa*.

#### 4. DISCUSSION and CONCLUSION

There are some data on the chemical composition of *C. spinosa* in the literature. According to these data, the presence of some tannins, saponins, alkaloids and flavonoids has been detected in this plant so far (Anwar *et al.*, 2016; Snoussi *et al.*, 2017). It is of course possible to elaborate on these studies. However, it is seen that some phytochemicals specific to this species come to the fore in some studies. In a study by Fu *et al.* (2007), cappariloside A and stachydrin were found to be the main components, while in some other studies, it was reported that rutin, which is a flavonoid, is found in high amounts in the plant (Stefanucci *et al.*, 2018; Mollica *et al.*, 2019). A review by Anwar *et al.* (2016) documented the flavonoids, alkaloids and essential oil components identified so far in *C. spinosa*. However, none of these studies included the presence of hepseridine and hyperoside, which were identified as the main compounds in the current study. Therefore, the presence of these compounds in *C. spinosa* was brought to the literature for the first time with this study.

**Figure 5.** The capacity of the extracts of *C. spinosa*. to inhibit some enzymes. Different letters (a, b, c) on the bars show that the relevant data are statistically different from each other (p < 0.05).

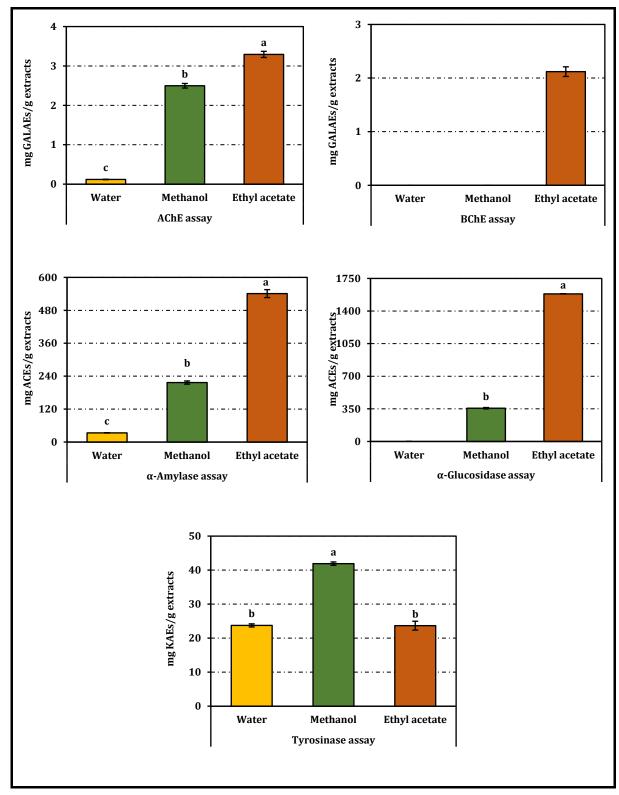


Table 2. Correlations among chemical composition and assays.

				C		•		•				
	22	1	2	3	4	5	7	8	9	10	11	
1	0.986											
2	0.989	0.999										
3	0.990	0.999	0.999									
4	0.987	0.999	0.999	0.999								
5	0.985	0.999	0.999	0.999	0.999							
6	0.988	0.999	0.999	0.999	0.999	0.999						
7	-0.976	-0.992	-0.991	-0.993	-0.991	-0.989						
8	-0.993	-0.999	-0.999	-0.999	-0.999	-0.999	0.987					
9	0.991	0.997	0.998	0.997	0.998	0.998	-0.981	-0.999				
10	-0.974	-0.998	-0.996	-0.996	-0.997	-0.997	0.996	0.993	-0.989			
11	-0.989	-0.999	-0.999	-0.999	-0.999	-0.999	0.993	0.999	-0.997	0.997		
12	0.989	0.999	0.999	0.999	0.999	0.999	-0.991	-0.999	0.998	-0.996	-0.999	
13	0.989	0.999	0.999	0.999	0.999	0.999	-0.993	-0.999	0.997	-0.997	-0.999	
14	0.986	0.999	0.999	0.999	0.999	0.999	-0.991	-0.999	0.997	-0.997	-0.999	
15	0.987	0.999	0.999	0.999	0.999	0.999	-0.993	-0.999	0.997	-0.997	-0.999	
16	0.980	0.997	0.997	0.995	0.997	0.998	-0.979	-0.996	0.997	-0.992	-0.995	
17	0.986	0.999	0.999	0.999	0.999	0.999	-0.992	-0.999	0.997	-0.997	-0.999	
18	0.989	0.999	0.999	0.999	0.999	0.998	-0.996	-0.998	0.995	-0.997	-0.999	
19	0.989	0.999	0.999	0.999	0.999	0.999	-0.993	-0.999	0.997	-0.997	-0.999	
20	0.989	0.999	0.999	0.999	0.999	0.999	-0.993	-0.999	0.997	-0.997	-0.999	
21	0.989	0.999	0.999	0.999	0.999	0.999	-0.993	-0.999	0.997	-0.997	-0.999	

1: DPPH, 2: ABTS, 3: CUPRAC, 4: FRAP, 5: FICA, 6: RACI: 7: AChEIA, 8: BChEIA, 9: TIA, 10: AAIA, 11: AGIA, 12: Total flavonoid, 13: Total phenolic, 14: Protocatechuic acid, 15: 4-Hydroxybenzoic acid, 16: Vanillic acid, 17: p-Coumaric acid, 18: Ferulic acid, 19: Hesperidin, 20: Hyperoside, 21: Quercetin, 22: TAP

As detailed in Part 3, the MeOH extract of *C. spinosa* showed remarkable antioxidant activity. There are many studies on the antioxidant activity of *C. spinosa* in the literature (Nadaroglu *et al.*, 2009; Tlili *et al.*, 2017; Yu *et al.*, 2017; Al-Azawi *et al.*, 2018). In many of these studies, the plant species in question exhibited remarkable antioxidant and radical scavenging activity. Therefore, the data obtained from the present study confirm the literature data. Also, according to the literature data, quercetin (Selway, 1986; Rauha *et al.*, 2000; Guardia *et al.*, 2001; Williams *et al.*, 2004), 4-hydroxybenzoic acid (Duke *et al.*, 2003; Manuja *et al.*, 2013) and *p*-coumaric acid (Bonina *et al.*, 2002; Ahmad *et al.*, 2006) obtained from this plant may be phytochemicals responsible for the antioxidant activity. However, as mentioned above, the presence of hesperidin and hyperoside in this plant was brought to the literature for the first time with this study. There are some literature data on the contribution of these compounds to antioxidant activity with some other plants or with these compounds themselves (Ku *et al.*, 2014; Hao *et al.*, 2016; Yatao *et al.*, 2018; Gao *et al.*, 2019; He *et al.*, 2019; Kim *et al.*, 2019; Musa *et al.*, 2019; Aggarwal *et al.*, 2020; Huang *et al.*, 2020). These findings support the correlation coefficient data obtained from the present study.

There are some reports in the literature regarding the cholinesterase inhibitory activity of *C. spinosa*. In a study carried out by Mollica *et al.* (2019), cholinesterase inhibitory activities of extracts obtained from *C. spinosa* by different methods were investigated and it was reported that the highest activity was exhibited by the extract obtained by microwave extraction. In another study by Wojdylo *et al.* (2019), it was reported that extracts obtained from different developmental stages of *C. spinosa*, especially those rich in flavonols (quercetin, kaempferol, myricetin, and isorhamnetin derivatives), showed significant cholinesterase inhibitory activity. Similar findings were also reported by Mekinic *et al.* (2018).

In the current study, the EtOAc extract exhibited the highest cholinesterase inhibitory activity, and according to the data in Table 1, this extract contains high amounts of protocatechuic and 4-hydroxybenzoic acids. There are some reports in the literature that these compounds themselves or some extracts containing high amounts of these compounds exhibit significant cholinesterase inhibitory activity (Szwajgier & Borowiec, 2012; Ertas *et al.*, 2014; Zengin *et al.*, 2017). These reports corroborate the data from the present study.

According to literature data, C. spinosa is considered to be a remarkable anti-hyperglycemic agent, in addition to its biological activities given above. In a study by Mollica  $et \ al. (2017)$ , it was reported that C. spinosa leaves or buds normalized biochemical parameters and reversed liver/lung damage in streptozocin-induced diabetic rats. The inhibitory activity of C. spinosa phytochemicals on  $\alpha$ -amylase and  $\alpha$ -glucosidase was also analyzed by  $in \ silico$  methods (Ogunwa  $et \ al., 2017$ ). In the aforementioned study, it was reported that naringin and rutin show high affinity for  $\alpha$ -amylase and  $\alpha$ -glucosidase. In the present study, EtOAc extract from C. spinosa exhibited the highest inhibitory activity on both enzymes. As can be seen from the data in Table 1, protocatechuic and 4-hydroxybenzoic acids are present in high amounts in this extract. Literature data indicate that both compounds may be responsible for the anti-diabetic activity of the extract (Saltan  $et \ al., 2017$ ; Alegbe  $et \ al., 2019$ ).

There are also some reports in the literature regarding the tyrosinase inhibitory activity of C. spinosa. It was determined that quercetin increased tyrosinase expression in B16 murine melanoma cells treated with C. spinosa extract at a concentration of 0.03% (w/v) (Matsuyama et al., 2009). Similar findings have been reported in a different report of the same research group (Matsuyama et al., 2009). In the current study, MeOH extract exhibited the highest tyrosinase inhibitory activity. According to the data in Table 1, it is thought that the main compounds of the MeOH extract contribute significantly to this activity. However, the presence of 1335.88  $\mu$ g/g quercetin in the MeOH extract creates a contradiction between the data obtained from the current study and the literature data. Therefore, biological activity-guided fractionation is needed to elucidate the compounds that contribute to the activity.

## **Declaration of Conflicting Interests and Ethics**

The authors declare no conflict of interest. This research study complies with research and publishing ethics. The scientific and legal responsibility for manuscripts published in IJSM belongs to the authors.

## **Authorship Contribution Statement**

Bulent Kirkan: Investigation and Resources. Olcay Ceylan: Resources. Cengiz Sarikurkcu: Methodology, Visualization, Software, Formal Analysis and Validation. Bektas Tepe: Investigation, Supervision and Writing -original draft.

### Orcid

Bulent Kirkan https://orcid.org/0000-0003-3462-0681
Olcay Ceylan https://orcid.org/0000-0002-4371-2126
Cengiz Sarikurkcu https://orcid.org/0000-0001-5094-2520
Bektas Tepe https://orcid.org/0000-0001-8982-5188

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